

Chapter 21

Stellar Explosions

Dr. Tariq Al-Abdullah



Learning Goals:

21.1 Life After Death for White Dwarf

21.2 The End of a High-Mass Star

21.3 Supernovae

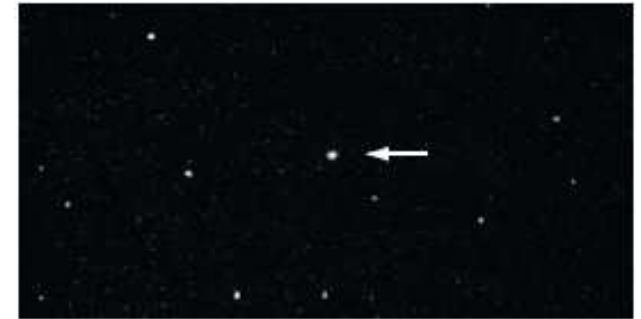
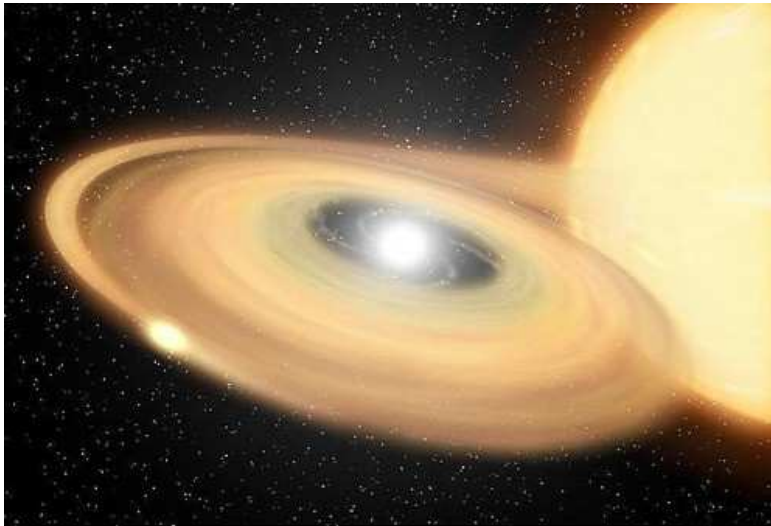
21.4 Formation of Elements

21.5 The Cycle of Stellar Evolution

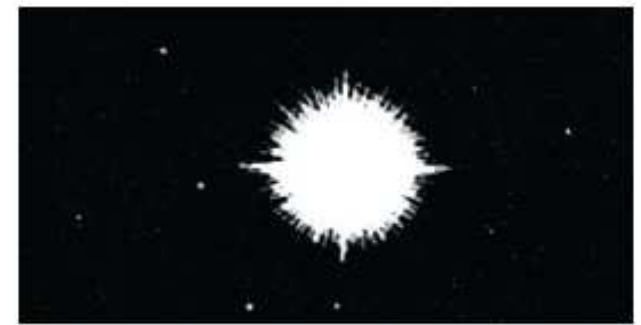
21.1 Life After Death for White Dwarfs

A nova: a new star that flares up very suddenly and then returns slowly to its former luminosity

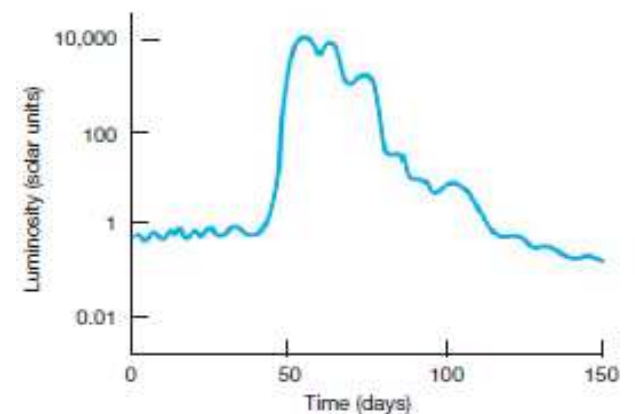
- Typically 2-3 novae are observed each year.
- Recurrent novae: stars that “go nova” several times.
- What cause such an explosive on a faint-dead star?
- A binary system gives the answer:



(a)



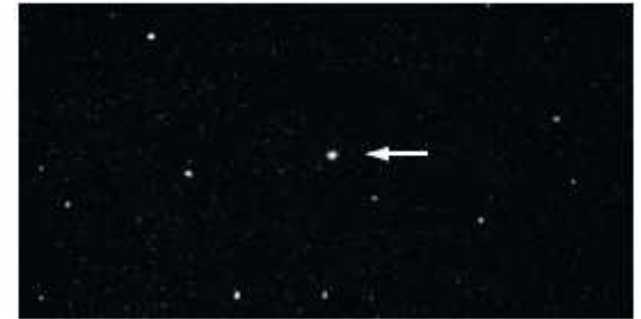
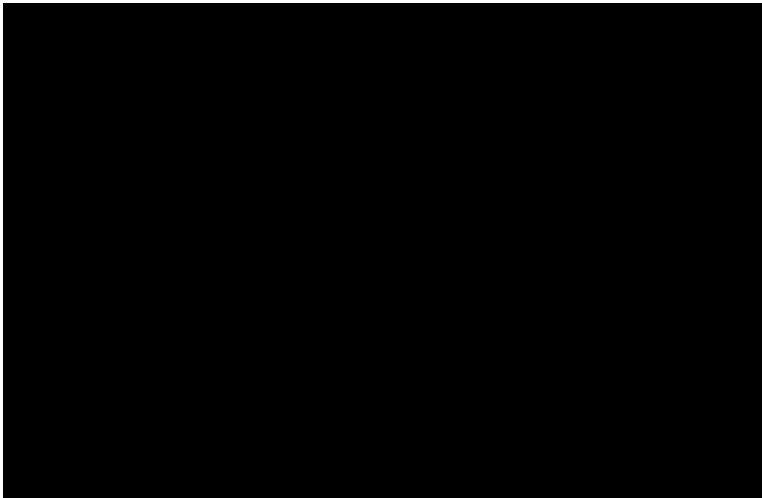
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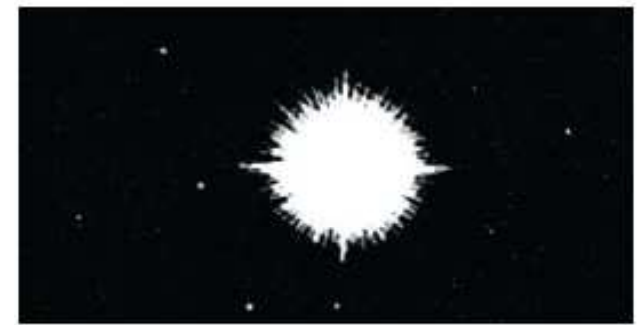
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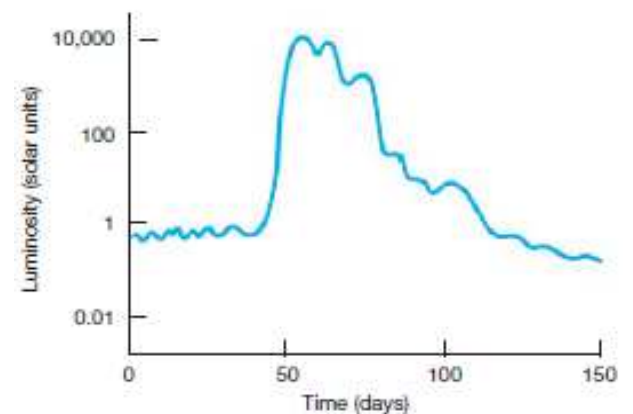
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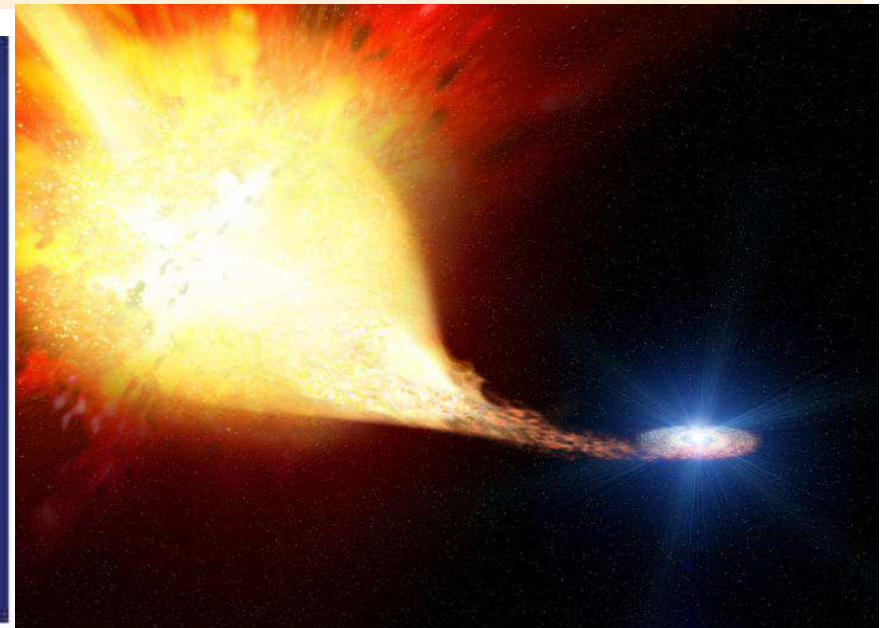
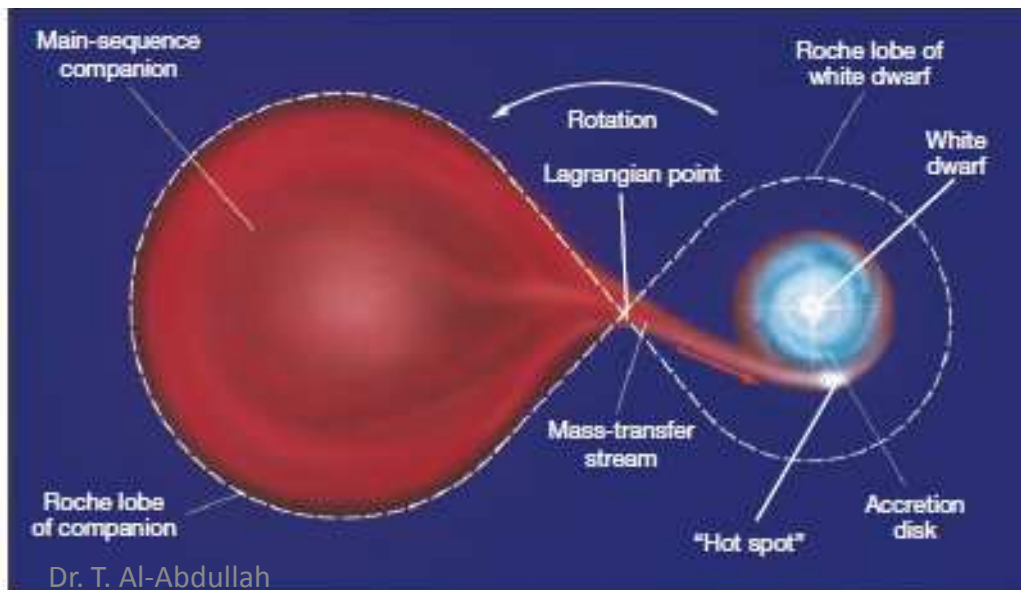
(b)



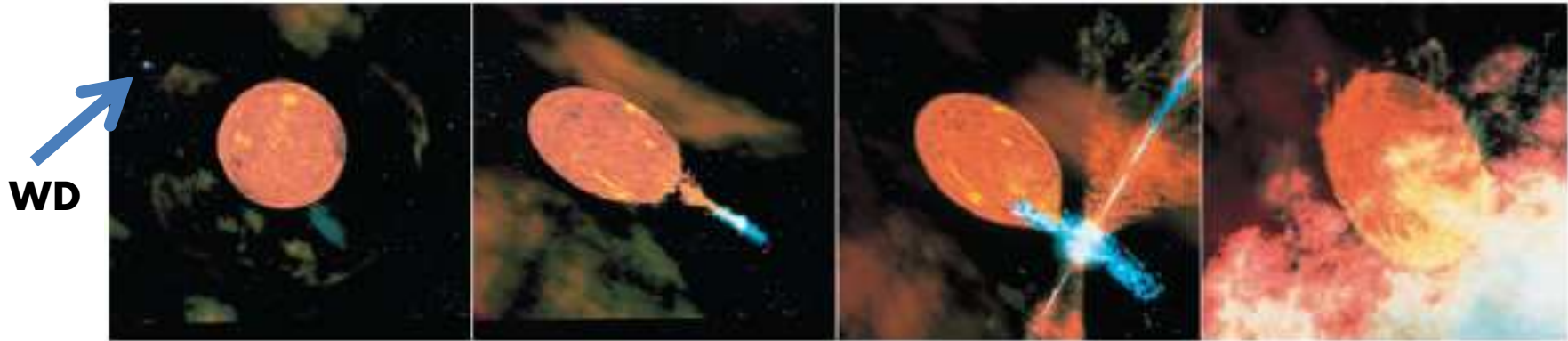
21.1 Life After Death for White Dwarfs

Accretion disk: missed materials from the giant companion loops around the white dwarf, goes into orbits around it forming swirling flattened disk of matter.

- Temperature on the outer surface increased $\rightarrow 10^7$ K, transferred gas denser.
- H fused into He at a fast rate \rightarrow surface burning is violent.
- The star suddenly flares up its luminosity and fades away.
- Part of the disk is blown off into space.



21.1 Life After Death for White Dwarfs



- Spectrum: visible, UV and X-rays
- Matter strikes the accretion disk form a turbulent “hotspots” → fluctuation in the emitted light.
- A nova’s decline in brightness due to cooling of its surface layers as they are blown in the space
- Two novae spelling mass from their surfaces.
- Extending the life of the white dwarf.



(a) Nova Persei, 1901, fig after 50y.



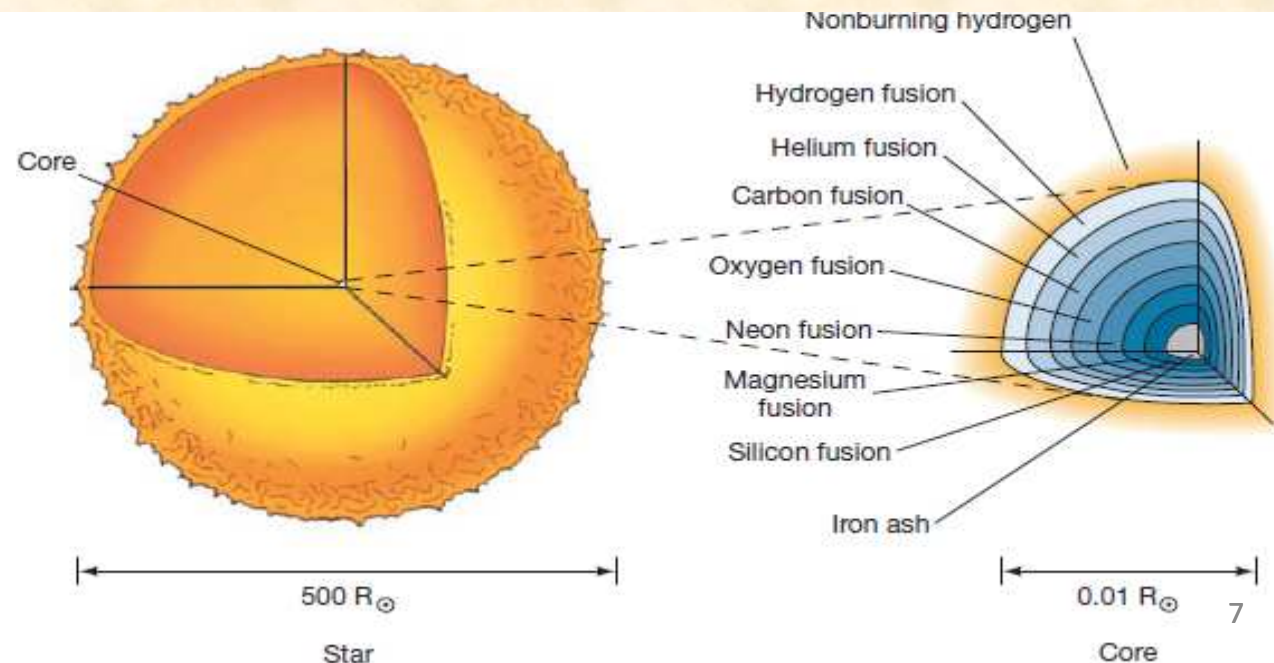
(b) Nova Cygni, 1992, ring after 7 months

21.2 The End of a High-Mass Star

→ Fusion of Heavy Elements $> 8 M_{\text{sun}}$??

- As the Temperature increases, the ash of the each burning stage becomes a fuel for the next stage.
- As each element is burned at the center, the core contracts, heats up, and start to fuse the ash of the previous stage → new element created.
- Inner core: Iron – Si – Mg – Ne – O – C – He – H – non-burning H

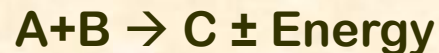
- Burning stages:
 - * H: 10 million years
 - * He: 1 million year
 - * C: 1000 year
 - * O: one year
 - * Si: one week
 - * Fe: < a day



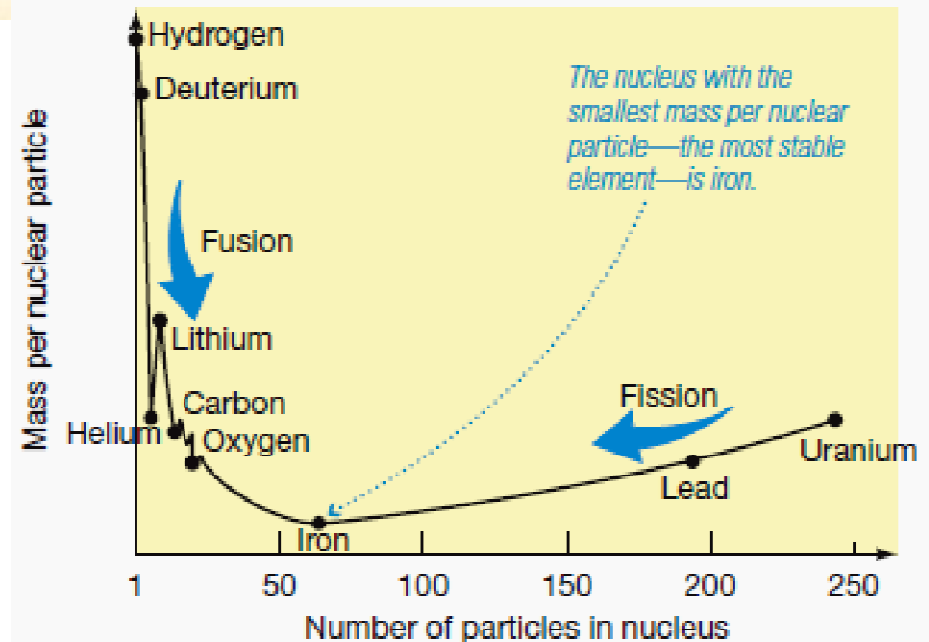
21.2 The End of a High-Mass Star

→ Collapse of the Iron Core

- Once the core is Fe → the high mass star is in trouble.
- Iron is the most stable element; ratio: mass / number of nucleons.
- Fusion versus, Fission reactions, both ends to Fe → central fires cease for the last time → destroy equilibrium forever.
- Enormous inward gravity >> thermal pressure → Star implodes on itself!!



- $M_A + M_B > M_C \rightarrow$ exothermic
- $M_A + M_B < M_C \rightarrow$ endothermic
- Triple-alpha up to ^{56}Ni decays ^{56}Fe



21.2 The End of a High-Mass Star

→ Collapse of the Iron Core

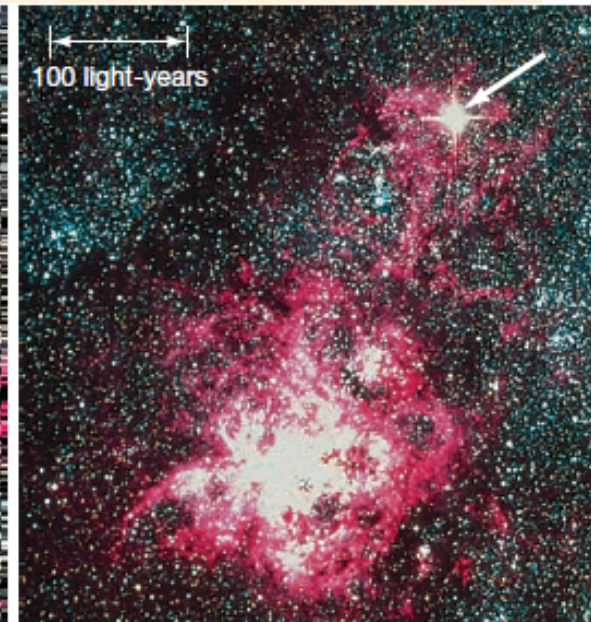
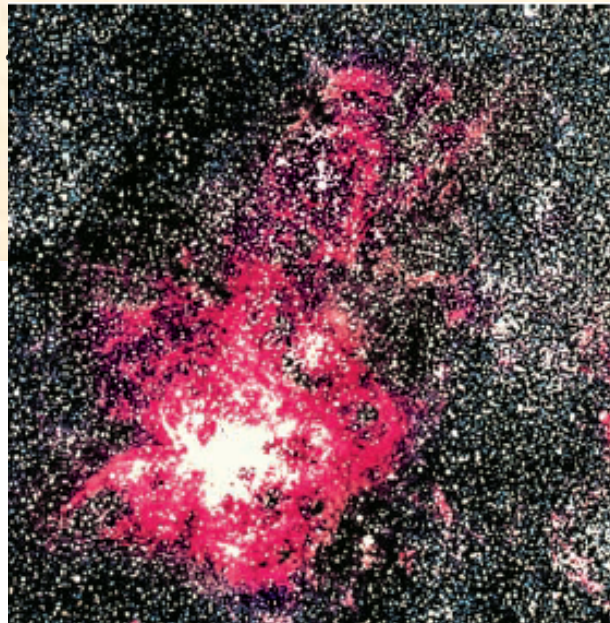
- The temperature rises to 10^{10} K → Wien's Law → photons have high Energy
- **Photo disintegration: Photons split heavy elements into lighter ones.**
- **10 million years to build Fe-core, 1 sec to destroy everything back to H & n.**
- The process absorbs thermal energy, not producing it → cooling the core.
- **Core unable to support itself against its own gravity → Accelerate core-collapsing**
- Core consists of elementary particles: protons, electrons, neutrons, photons.
- **Because of the high density: $p + e \rightarrow n + \text{neutrino}$, neutrinos don't support the thermal pressure → reducing the core's pressure support.**
- This process is called: **neutronization of the core.**
- **Density now: 10^{12} kg/m³.**

21.2 The End of a High-Mass Star

→ Supernova Explosion

- Disappearance of electrons and escaping the neutrinos → nothing to prevent the collapsing all the way to the point where neutron are in contact.
- Density 10^{15} kg/m³. Neutron slow compression, density up to $10^{17,18}$ kg/m³
- Rebounding process, expansion again → Shock Wave
- Blasting all the overlying layers, including heavy elements.
- Computer models uncertain
- Most energetic events;
- Core-collapse supernova

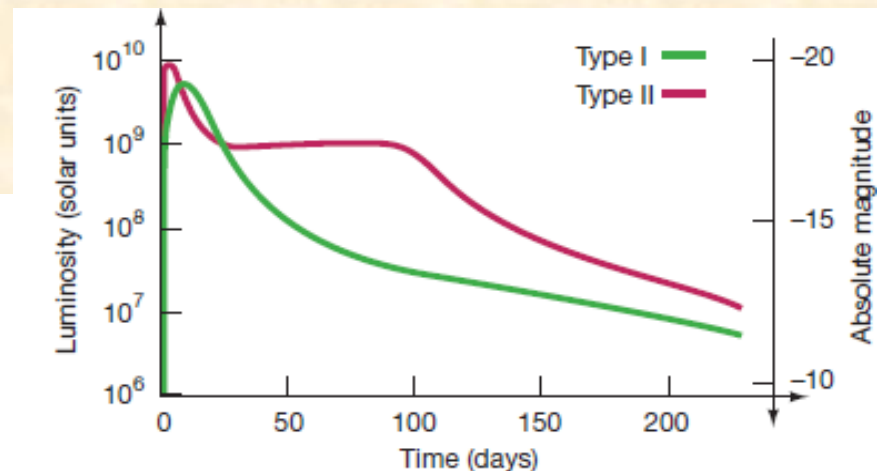
SN 1987A



21.3 Supernovae

Novae and Supernovae

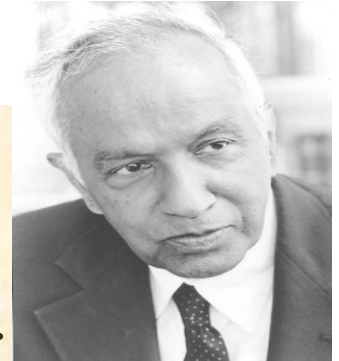
- Supernova is million times brighter than nova
- Energy radiated by supernova in few months, 10^{43} J = Energy radiated by the Sun in 10 billion years.
- Same star may become a nova several times, but becomes a supernova only once.
- Supernova contain very little H, while others contain no H
- Type I supernova: H-poor kind, light curve similar to nova
- Type II Supernova: H-rich kind
- SN is either Type I or Type II.



21.3 Supernovae

Carbon-Detonation Supernovae (SN-I)

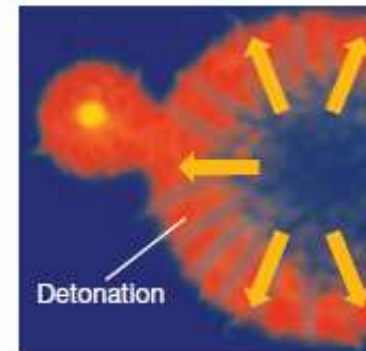
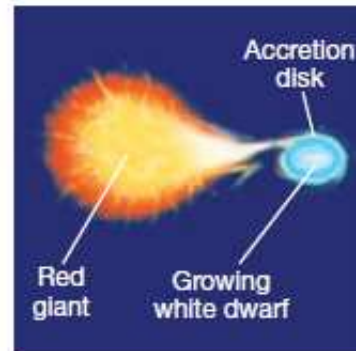
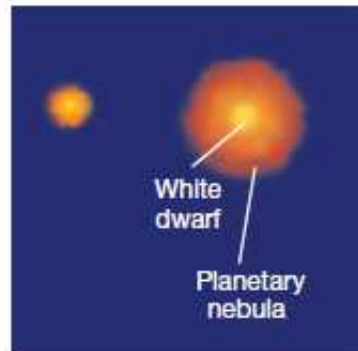
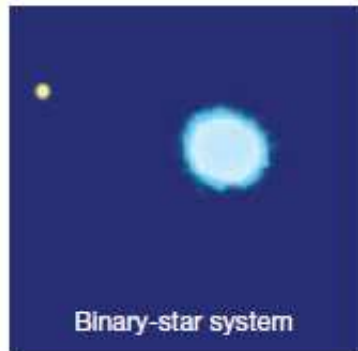
- *What is the responsible for the differences among supernovae?*
- White dwarf maximum mass limit is $1.4M_{\text{sun}}$, Chandrasekhar mass.
- White dwarf: supported by electron-degeneracy pressure against the gravity
- Its mass, not all ejected, grows from the companion → new period of instability.
- If $M > 1.4M_{\text{sun}}$, the gravity wins → collapsing & heating → Fusing Carbon everywhere.
- The whole star explode → Carbon-Detonation supernova
- Two white dwarfs may emerged → massive unstable star → Type I SN
- SN-I origin is white dwarf → No H. Light curve results from the radioactive decay of unstable heavy elements.
- Implosion-explosion of SN produces Type II. Unburned H and He.



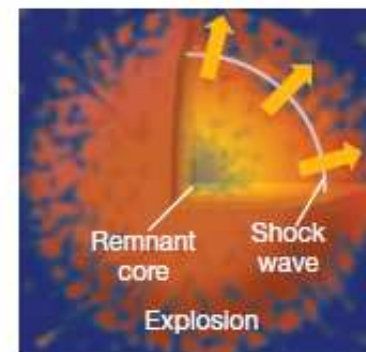
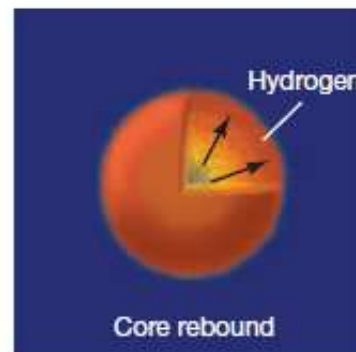
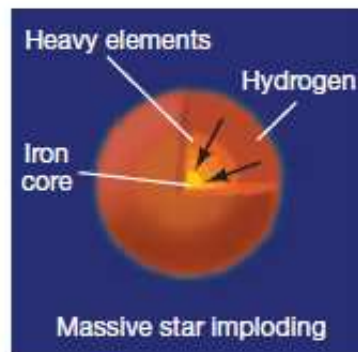
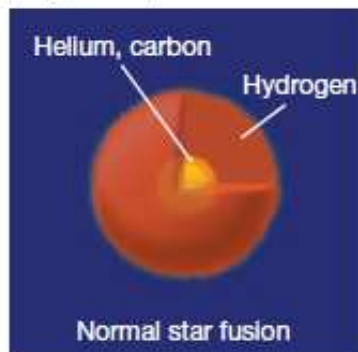
21.3 Supernovae

Two-types of supernova occur at roughly the same rate.

(a) Type I Supernova

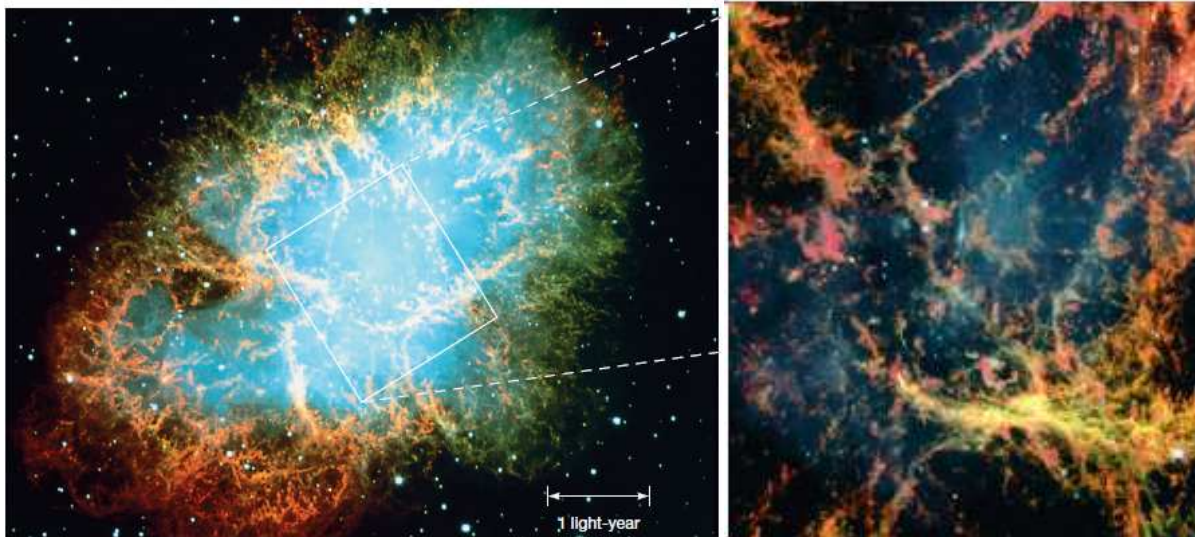


(b) Type II Supernova



Supernova Remnants

- *Several evidences that SN occurs.*
- Their glowing remains (Remnants).
- **Example: Crab Nebula was huge SN in 1054 A.D. Chinese and Middle Eastern astronomers.**
- **Was as large as Venus, some said as moon for almost one month.**
- Mass ejected from central explosion → Type II, expanding into space at several thousands km/s.

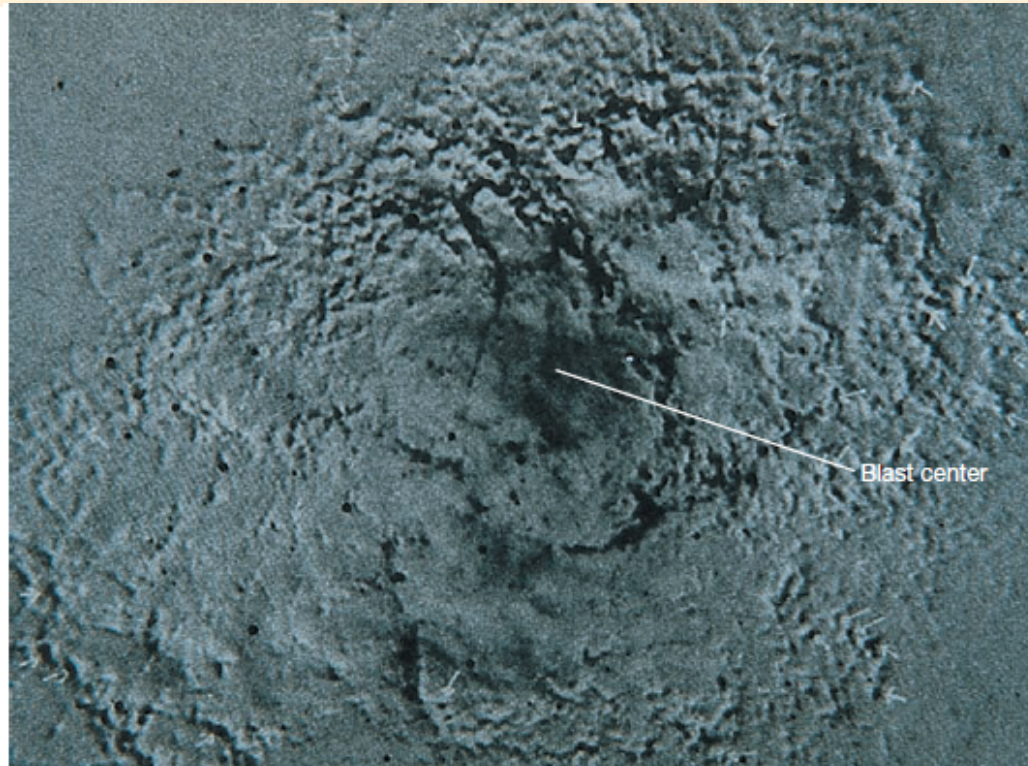


21.3 Supernovae

Supernova Remnants

Vela SN: Expansion velocities imply that its center explode 9000 B.C.

- Remnant is only 1600 ly from us, was as bright as moon for several months.
- Stone-Aged humans



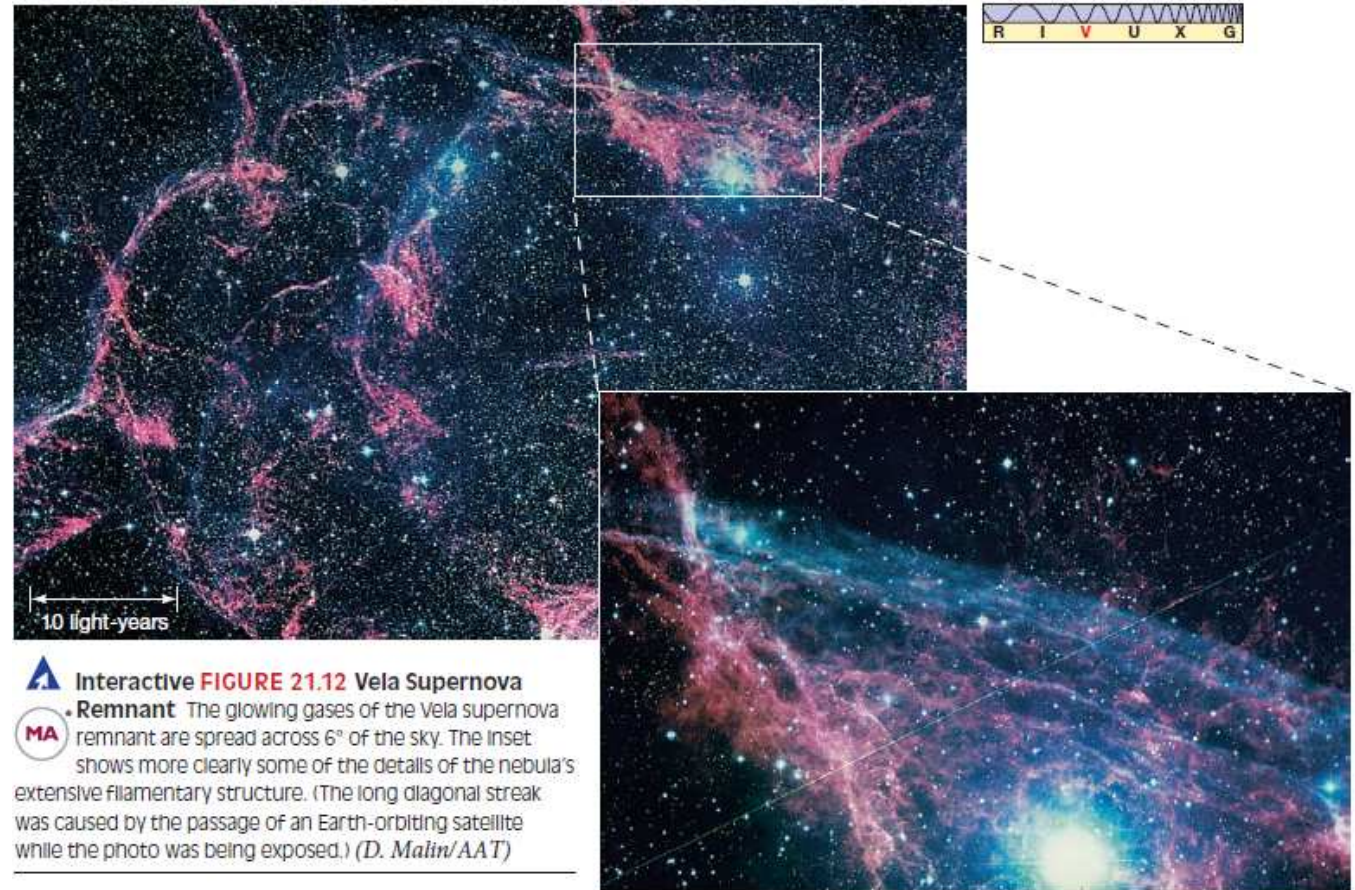
21.3 Supernovae

Vela SN now

Supernova Remnants

No SN in Milky-Way galaxy has been observed since Tycho (1572) and Kepler (1604).

- Expected: 1/100 yr



21.4 Formation of the Elements

Types of Matter:

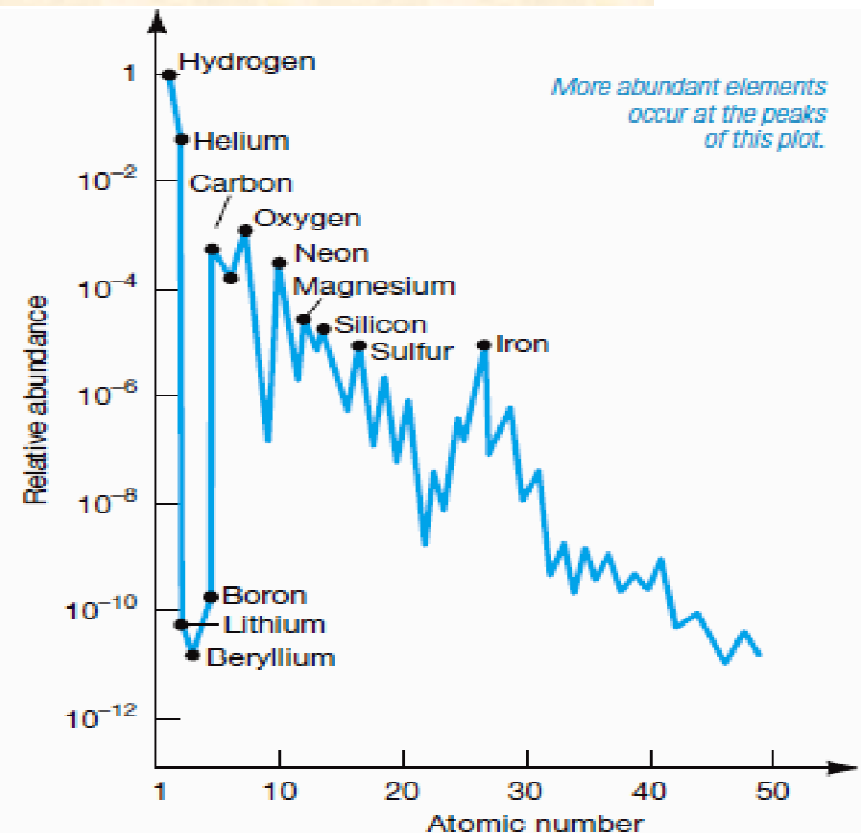
- 115 elements : H (1p) → Ununpentium (115p & 184n) are known.
- 81 stable elements + 10 radioactive elements found on Earth.

• Radon and Uranium exist long time ago, very long half-life times.

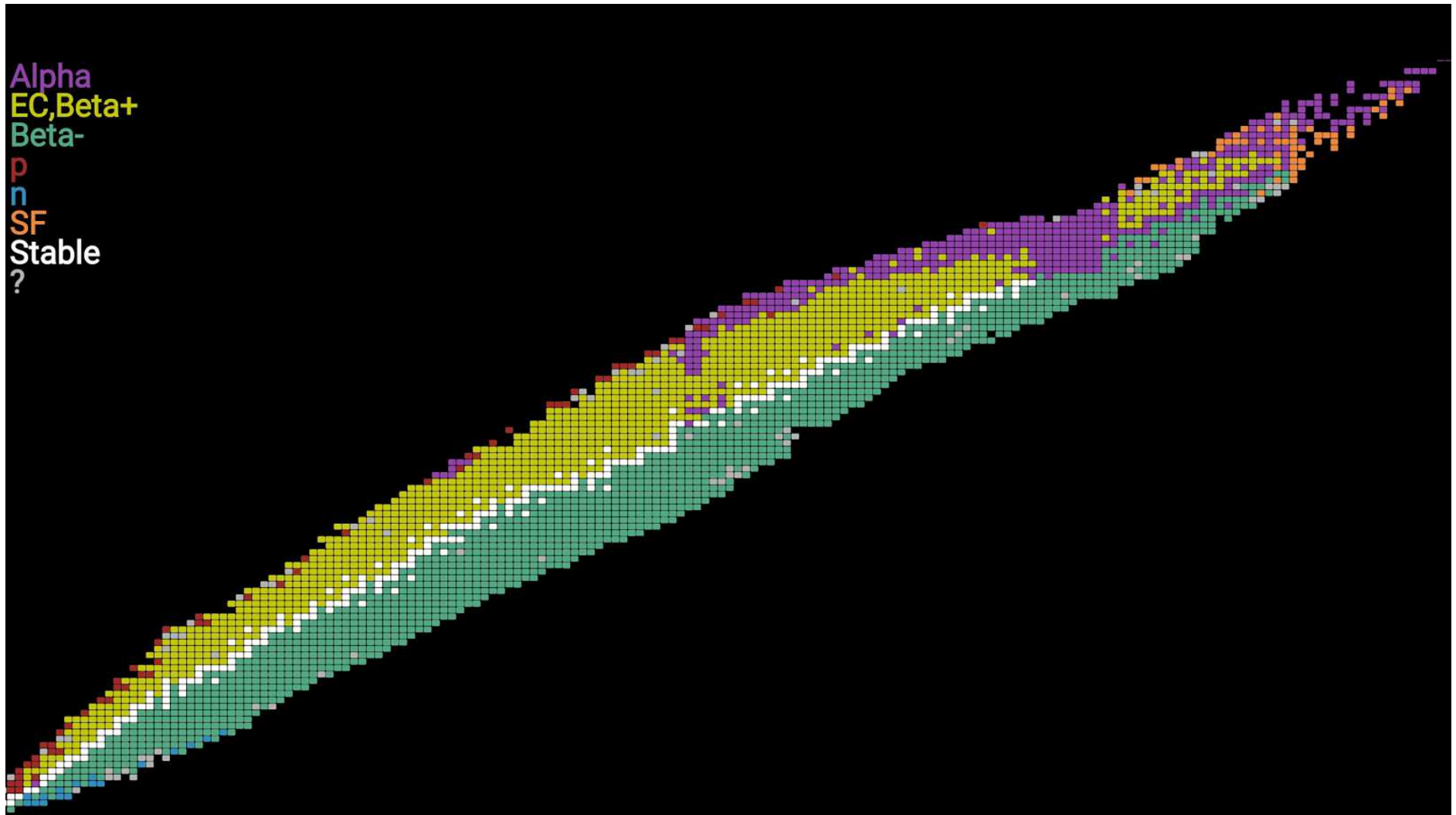
• 19 radioactive elements produced artificially in nuclear laboratories.

PERIODIC TABLE OF THE ELEMENTS

The periodic table shows elements from Hydrogen (H) to Oganesson (Og). It is divided into several categories: Nonmetals, Metals, and Metalloids. Specific groups are labeled: Group 1A (Alkali metals), Group 2A (Alkaline earth metals), Group 7A (Halogens), and Group 8A (Noble gases). The table also includes the Lanthanide and Actinide series at the bottom.



- All elements exist in different isotopes. Thousands of isotopes



21.4 Formation of the Elements

Abundance of Matter: Stellar nucleosynthesis.

TABLE 21.1 Cosmic Abundances of the Elements

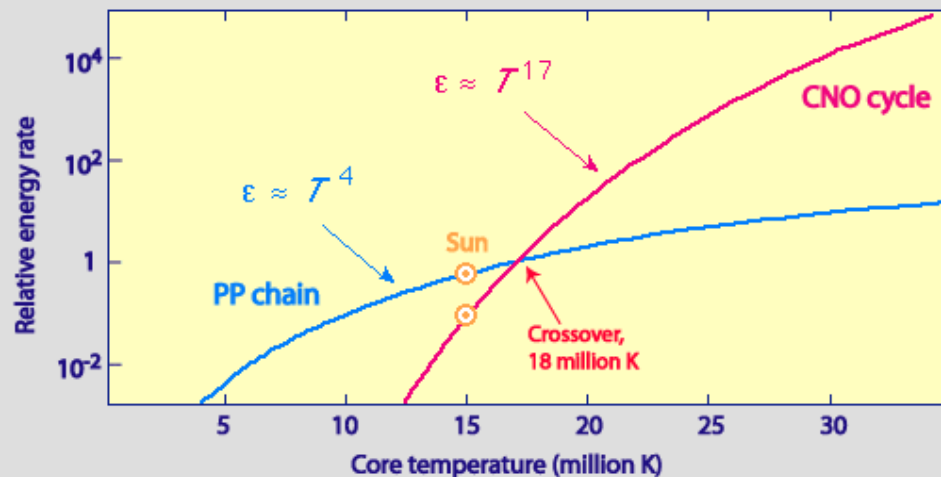
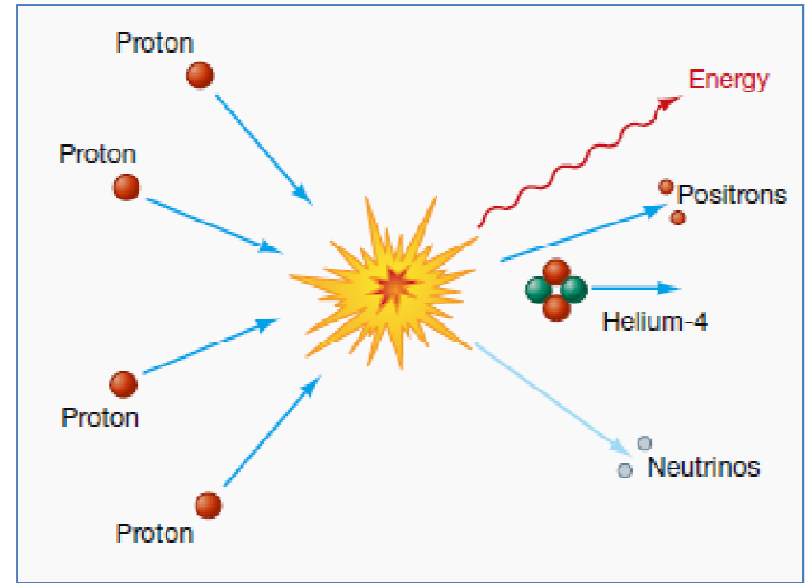
Elemental Group of Particles	Percent Abundance by Number*
Hydrogen (1 nuclear particle)	90
Helium (4 nuclear particles)	9
Lithium group (7–11 nuclear particles)	0.000001
Carbon group (12–20 nuclear particles)	0.2
Silicon group (23–48 nuclear particles)	0.01
Iron group (50–62 nuclear particles)	0.01
Middle-weight group (63–100 nuclear particles)	0.00000001
Heaviest-weight group (over 100 nuclear particles)	0.000000001

** The total does not equal 100 percent because of uncertainties in the abundance of helium. All isotopes of all elements are included.*

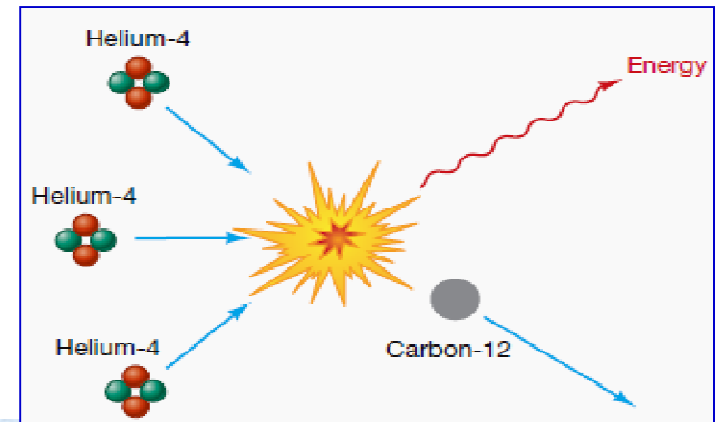
H & He burning

10 million K, pp-cycle

- $4(^1\text{H}) \rightarrow ^4\text{He} + 2\text{e}^+ + 2\nu + \text{energy}$
- e^+ annihilate directly with $\text{e}^- \rightarrow \gamma = 511 \text{ keV}$
- neutrinos rapidly escape, play no role.
- 20 million K (accelerate the H-burning)
- In massive stars: CNO takes over.



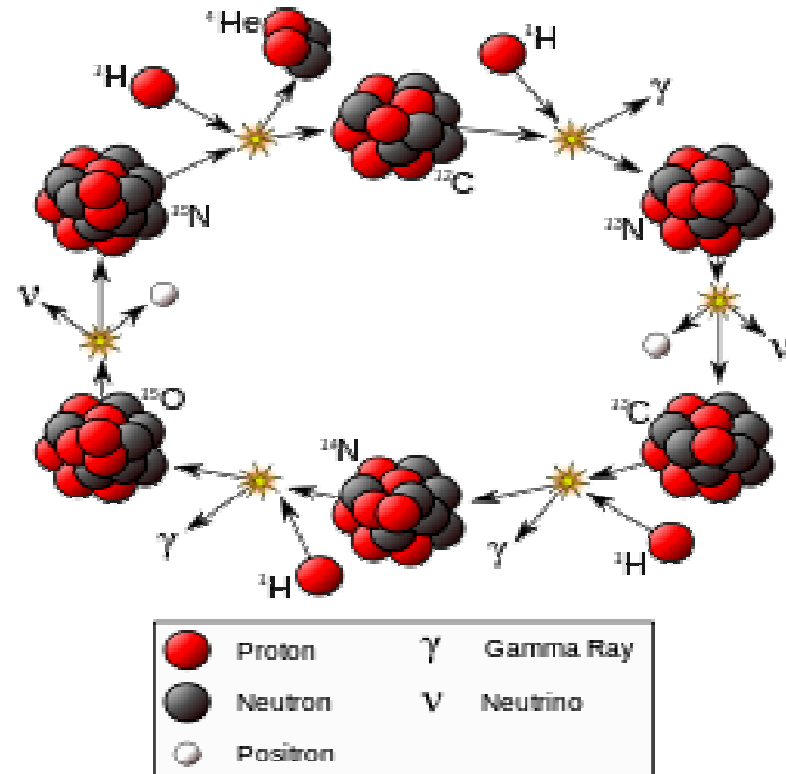
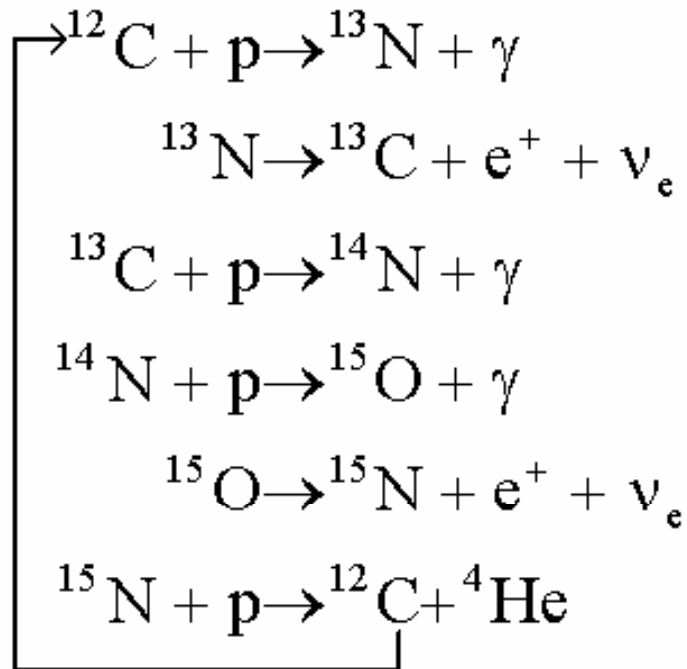
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100 million K, triple-alpha process



CNO cycle



The abundance of carbon, nitrogen, and oxygen is unchanged.

Simply H nuclei are combined to form helium.

Same result as: $4({}^1\text{H}) \rightarrow {}^4\text{He} + 2 \text{e}^+ + 2 \nu + \text{energy}$.

21.4 Formation of the Elements

C burning & He Capture

600 million K,

-- $^{12}\text{C} + ^{12}\text{C} \rightarrow ^{24}\text{Mg} + \text{energy}$, but

-- $^{12}\text{C} + ^4\text{He} \rightarrow ^{16}\text{O} + \text{energy}$, more probable

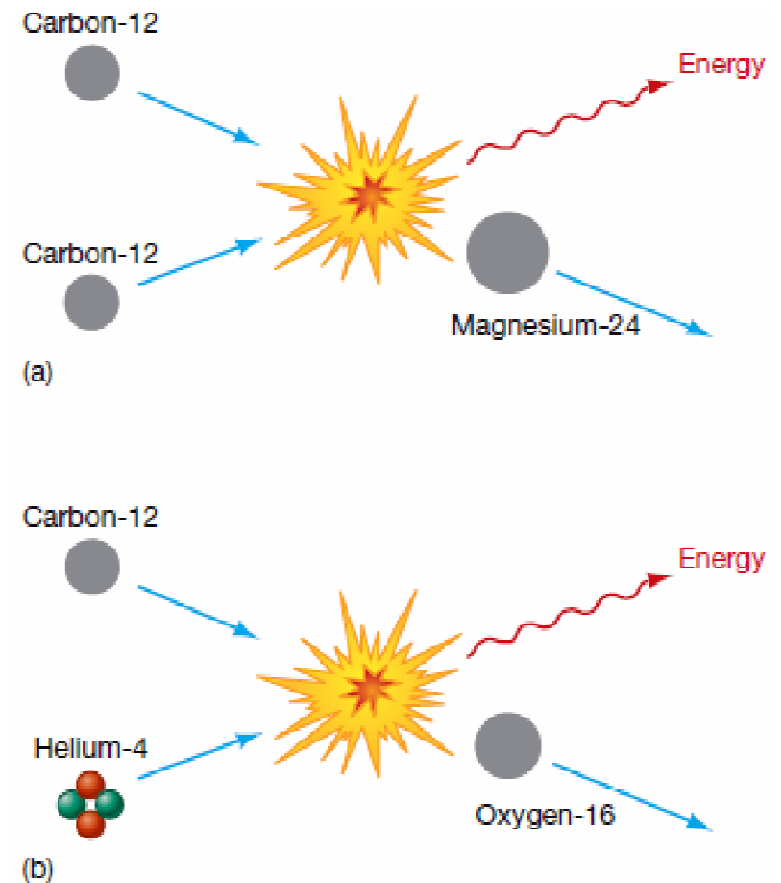
-- Similarly:

-- $^{16}\text{O} + ^{16}\text{O} \rightarrow ^{32}\text{S} + \text{energy}$, but

-- $^{16}\text{O} + ^4\text{He} \rightarrow ^{20}\text{Ne} + \text{energy}$, more probable

➔ He-capture process, 4(He)–12(C)–16(O)–20(Ne)–24(Mg)–28(Si) are prominent nuclei.

➔ Other nuclei might be produced in similar way such as: ^{19}F , ^{23}Na , ^{31}P , low abundance.



21.4 Formation of the Elements

Iron Formation (alpha process)

3 billion K

-- Once ^{28}Si appears in the core \rightarrow struggling begins between He-capture or breaking complex nuclei due to heat (photodisintegration).

\rightarrow 2-step process: photodisintegration then He-capture

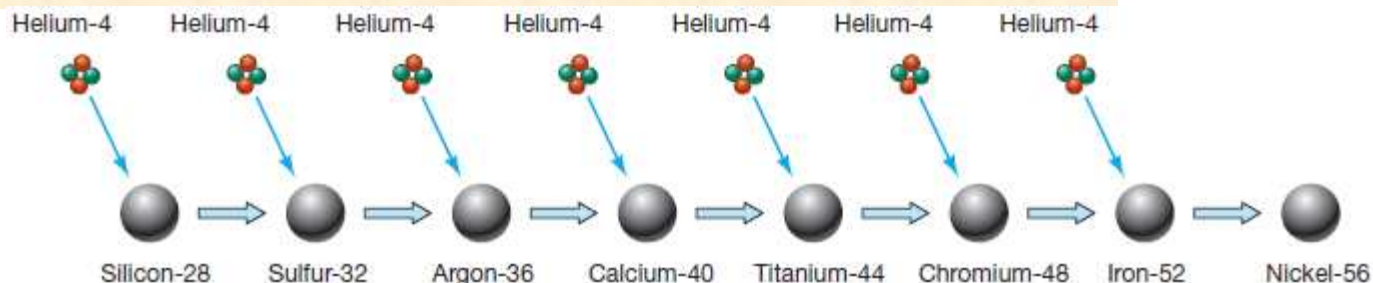
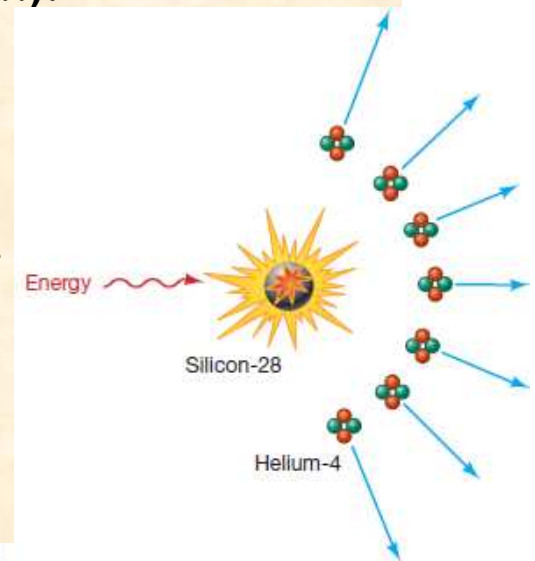
-- ^{28}Si breaks into seven ^4He nuclei.

-- Star forms: ^{32}S , ^{36}Ar , ^{40}Ca , ^{44}Ti , ^{48}Cr , ^{52}Fe , ^{56}Ni and others.

-- Summary of the chain reaction:

-- $^{28}\text{Si} + 7(^4\text{He}) \rightarrow ^{56}\text{Ni} + \text{energy}$

$\rightarrow ^{56}\text{Ni} \rightarrow \beta\text{-decay} \rightarrow ^{56}\text{Co} \rightarrow \beta\text{-decay} \rightarrow ^{56}\text{Fe}$ (most stable)



Elements beyond Fe

Heavier elements are produced through neutron-capture reactions

-- neutron available “by-products” of nuclear reactions.

-- neutron easily interact with matter?

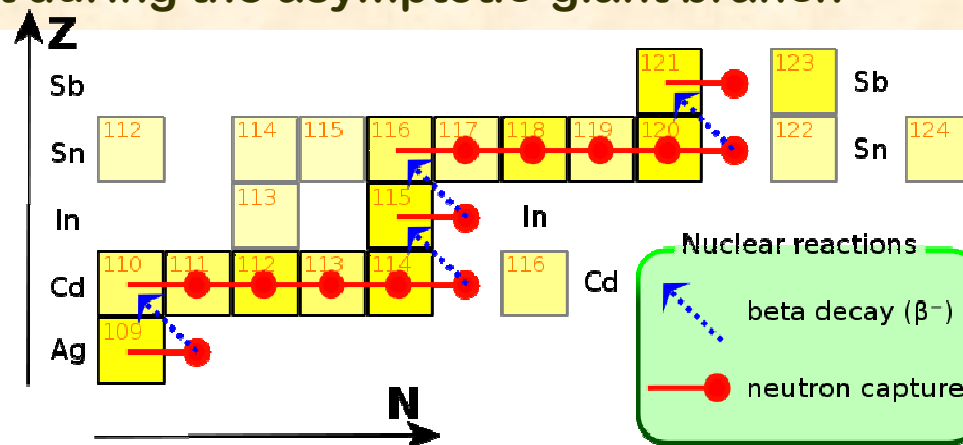
-- $^{56}\text{Fe} + n \rightarrow ^{57}\text{Fe} \rightarrow ^{57}\text{Fe} + n \rightarrow ^{58}\text{Fe} \rightarrow ^{58}\text{Fe} + n \rightarrow ^{59}\text{Fe}$

-- ^{59}Fe radioactive \rightarrow one month \rightarrow ^{59}Co stable

-- $^{59}\text{Co} + n \rightarrow ^{60}\text{Co}$ (unstable) decay to ^{60}Ni

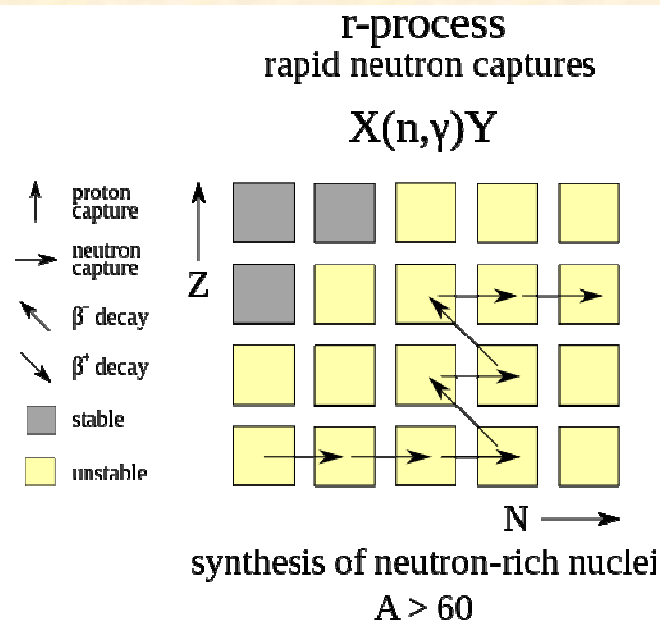
→ Neutron capture: s-process (slow) needs time (a year to capture n)

→ Are important during the asymptotic-giant branch



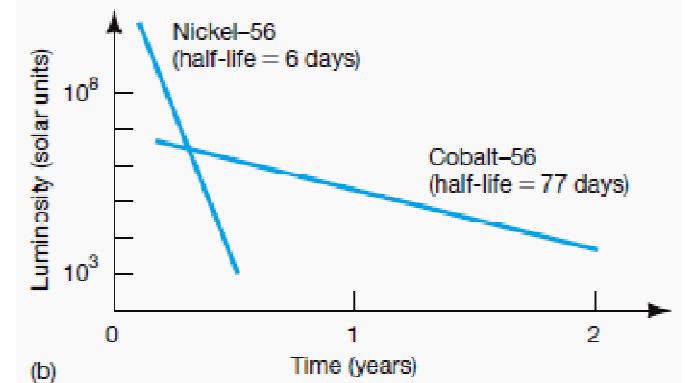
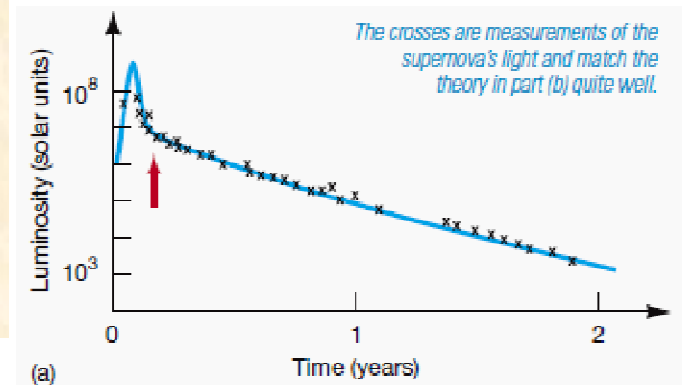
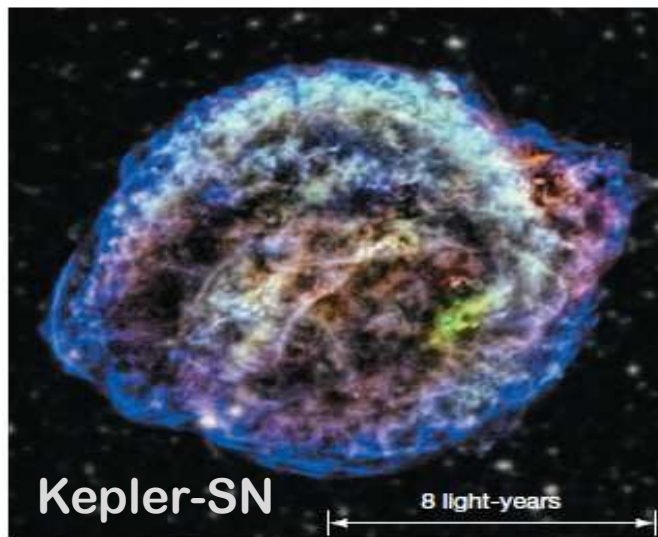
Making the Heaviest Elements

- s-process: stable nuclei up to ^{209}Bi , not for radioactive nuclei: ^{232}Th , ^{238}U , ^{242}Pu .
- Decay as fast as they produce by s-process, or even before
- r-process (rapid) occurs during supernova explosion.
- Free neutrons increase, fast enough for unstable to capture before decaying.
- Responsible to produce the heaviest elements.



Observational Evidences

- 1st: comparing nuclear reaction rates with observations
- 2nd: Technetium-99 ($T_{1/2} = 200,000$ yr) provides evidence that heavy elements form during SN.
- 3rd: Studying light-curve from SN-I, indicate which element is produced as a result of the explosion.



21.5 The Cycle of Stellar Evolution



END OF CHAPTER 21



Dr. Tariq Al-Abdullah