Phys 741 Statistical Mechanics Problem Set # 1

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1. Starting from dE = dQ - PdV and for an adiabatic process show that

$$C_P \left(\frac{\partial T}{\partial V}\right)_P dV + C_V \left(\frac{\partial T}{\partial P}\right)_V dP = 0$$

and verify that for an ideal gas this reduces to $PV^{\gamma} = const$, where $\gamma = \frac{C_P}{C_V}$ 2. Show that

$$\left(\frac{\partial C_P}{\partial P}\right)_T = -T \left(\frac{\partial^2 V}{\partial T^2}\right)_P$$

and

$$\left(\frac{\partial T}{\partial P}\right)_S = \left(T/C_P\right) \left(\frac{\partial V}{\partial T}\right)_P$$

- 3. Pathria 1.2
- 4. (a) Starting form dE = TdS PdV, show that for an ideal gas, the internal energy depends only on T, i.e. E = E(T)
 - (b) Write down a general expression for the equation of state that is consistent with E = E(T)
 - (c) Show that for a van der Waals gas C_V is a function of temperature alone
- 5. Three distinguishable particles can each occupy energy levels 0, ε , and 3ε . Calculate the entropy of the system if the total energy is $a E = 2\varepsilon$, $b E = 3\varepsilon$, $c E = 9\varepsilon$
- 6. Consider a particle of mass m moving freely in a 1D box of size L. Find the eigenvalues and the normalized eigenvectors $\phi_n(x)$. Now consider a system of 3 non-interacting particles of mass m placed in such 1D box. If the energy of the system equals to $E = 38\varepsilon_0$, where $\varepsilon_0 = \frac{\hbar^2 \pi^2}{2mL^2}$ is the lowest eigenvalue, find the entropy of the following three cases where the system is composed of:
 - (a) 3 distinguishable spinless particles
 - (b) 3 indistinguishable spinless bosons
 - (c) 3 indistinguishable spin-1/2 fermions

Good Luck