

# Graduate stat. Mech

## HW # 1 - solution

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$$\textcircled{1} \quad dE = dQ - PdV \Rightarrow dQ = dE + PdV$$

for adiabatic process  $dQ = 0 \Rightarrow 0 = dE + PdV \dots (1)$

equation of state contains only two independent variables.

let us pick up  $(P, V)$  i.e  $E = E(P, V)$

$$dE = \left(\frac{\partial E}{\partial P}\right)_V dP + \left(\frac{\partial E}{\partial V}\right)_P dV, \text{ substitute in (1), we get}$$

$$\left(\frac{\partial E}{\partial P}\right)_V dP + \left(\frac{\partial E}{\partial V}\right)_P dV + PdV = 0$$

$$\left[ \left(\frac{\partial E}{\partial V}\right)_P + P \right] dV + \left(\frac{\partial E}{\partial P}\right)_V dP = 0 \dots (2)$$

now  $\left(\frac{\partial E}{\partial V}\right)_P = \left(\frac{\partial E}{\partial T}\right)_P \left(\frac{\partial T}{\partial V}\right)_P$  chain rule  $\dots (3)$

and using  $dE = Tds - PdV$   $\begin{matrix} \text{1st and 2nd laws} \\ \text{0 adiabatic} \end{matrix}$

$$\begin{aligned} \left(\frac{\partial E}{\partial T}\right)_P &= T \left(\frac{\partial S}{\partial T}\right)_P + \cancel{ds} - P \left(\frac{\partial V}{\partial T}\right)_P - \left(\frac{\partial P}{\partial T}\right)_P \left(\frac{\partial V}{\partial T}\right)_P \\ &= C_p - P \left(\frac{\partial V}{\partial T}\right)_P \end{aligned}$$

eq<sup>n</sup> (3) becomes  $\left(\frac{\partial E}{\partial V}\right)_P = C_p \left(\frac{\partial T}{\partial V}\right)_P - P \left(\frac{\partial V}{\partial T}\right)_P \left(\frac{\partial T}{\partial V}\right)_P = C_p \left(\frac{\partial T}{\partial V}\right)_P - P$   $\dots (4)$

also from  $\left(\frac{\partial E}{\partial P}\right)_V = \left(\frac{\partial E}{\partial T}\right)_V \left(\frac{\partial T}{\partial P}\right)_P$  chain rule

$$= C_V \left(\frac{\partial T}{\partial P}\right)_P \quad \dots (5)$$

substitute 4 and 5 in 2, we get

$$C_P \left(\frac{\partial T}{\partial V}\right)_P dV + C_V \left(\frac{\partial T}{\partial P}\right)_V dP = 0 \quad \text{Q. 5. D}$$

- for an ideal gas  $PV = Nk_B T \Rightarrow T = \frac{PV}{Nk_B}$

$$\Rightarrow C_P \frac{dV}{V} + C_V \frac{dP}{P} = 0$$

integrate  $C_P \ln V + C_V \ln P = \text{const}$

divide by  $C_V \Rightarrow \frac{C_P}{C_V} \ln V + \ln P = \text{const}$

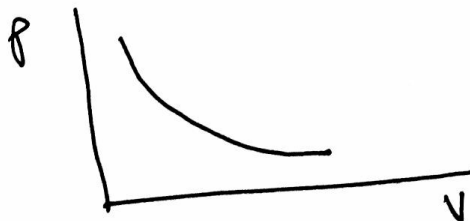
$$\delta \ln V + \ln P = \text{const}$$

$$\delta = \frac{C_P}{C_V}$$

$$\ln V^\delta + \ln P = \text{const}$$

$$\ln PV^\delta = \text{const}$$

$$\Rightarrow PV^\delta = \text{constant} \quad \underline{\text{Q. 5. D}}$$



②

$$a) \quad c_p = T \left( \frac{\partial S}{\partial T} \right)_p ; \quad \left( \frac{\partial c_p}{\partial p} \right)_T = T \frac{\partial}{\partial p} \left( \frac{\partial S}{\partial T} \right)_p$$

$$= T \frac{\partial}{\partial T} \left( \frac{\partial S}{\partial p} \right)_T \quad \dots (1)$$

but from

$$dG = -SdT + vdp + \mu dN \Rightarrow - \left( \frac{\partial S}{\partial p} \right)_T = \left( \frac{\partial v}{\partial T} \right)_p$$

$\Rightarrow$  eq<sup>n</sup> (1) becomes

Maxwell's relation

$$\left( \frac{\partial c_p}{\partial p} \right)_T = T \frac{\partial}{\partial T} \left( - \frac{\partial v}{\partial T} \right)_p = - T \left( \frac{\partial^2 v}{\partial T^2} \right)_p \quad \text{Q.E.D}$$

$$b) \quad \left( \frac{\partial T}{\partial p} \right)_s = \frac{\partial(T, S)}{\partial(p, S)} = \frac{\partial(T, S)}{\partial(T, p)} \frac{\partial(T, p)}{\partial(p, S)} = - \underbrace{\left( \frac{\partial S}{\partial p} \right)_T}_{?} \underbrace{\left( \frac{\partial T}{\partial S} \right)_p}_{?} \quad \dots (2)$$

from part a) we found  $\left( \frac{\partial S}{\partial p} \right)_T = - \left( \frac{\partial v}{\partial T} \right)_p$

$$\text{and using } \left( \frac{\partial T}{\partial S} \right)_p \left( \frac{\partial S}{\partial p} \right)_T \left( \frac{\partial p}{\partial T} \right)_s = -1 \Rightarrow \left( \frac{\partial T}{\partial S} \right)_p = - \frac{1}{\underbrace{\left( \frac{\partial S}{\partial p} \right)_T \left( \frac{\partial p}{\partial T} \right)_s}} \quad \dots (3)$$

$$\text{now } \left( \frac{\partial S}{\partial p} \right)_T \left( \frac{\partial p}{\partial T} \right)_s = \frac{\cancel{\partial(T, S)}}{\partial(T, p)} \frac{\partial(S, p)}{\cancel{\partial(S, T)}} = \left( \frac{\partial S}{\partial T} \right)_p = \frac{c_p}{T}$$

$$\Rightarrow \text{from (3)} \quad \left( \frac{\partial T}{\partial S} \right)_p = - \frac{1}{\left( \frac{\partial S}{\partial p} \right)_T} = - \frac{T}{c_p}$$

eq<sup>n</sup> (2) becomes

$$\left( \frac{\partial T}{\partial p} \right)_s = - \left( \frac{\partial v}{\partial T} \right)_p \left( - \frac{T}{c_p} \right) = \frac{T}{c_p} \left( \frac{\partial v}{\partial T} \right)_p \quad \text{Q.E.D}$$

(3)

③ problem 1.2

$$\text{given } s = s_1 + s_2 ; \quad \Omega = \Omega_1 \Omega_2 ; \quad s_1 = f(\Omega_1)$$

$$\text{show that } f(\Omega) = k \ln \Omega$$

$$s_2 = f(\Omega_2)$$

$$s = f(\Omega) = f(\Omega_1, \Omega_2)$$

starting from

$$s = s_1 + s_2$$

$$s = f(\Omega) = f(\Omega_1, \Omega_2) = f(\Omega_1) + f(\Omega_2)$$

$$\text{now } \frac{ds}{d\Omega_1} = \Omega_2 f'(\Omega_1, \Omega_2) = f'(\Omega_1) + \cancel{\frac{df(\Omega_2)}{d\Omega_1}} \rightarrow 0$$

$$\text{and } \frac{ds}{d\Omega_2} = \Omega_1 f'(\Omega_1, \Omega_2) = \cancel{\frac{df(\Omega_1)}{d\Omega_2}} + f'(\Omega_2) \rightarrow 0$$

$$\Rightarrow f'(\Omega_1, \Omega_2) = \frac{f'(\Omega_1)}{\Omega_2} = \frac{f'(\Omega_2)}{\Omega_1}$$

$$\Rightarrow \Omega_1 f'(\Omega_1) = \Omega_2 f'(\Omega_2)$$

L.H.S and R.H.S are functions of two different variables, so the only way they are equal is that both sides equal the same constant (say  $k$ )

$$\Rightarrow \Omega_1 f'(\Omega_1) = k \Rightarrow f'(\Omega_1) = \frac{k}{\Omega_1}$$

$$\text{integrate } f(\Omega_1) = k \ln \Omega_1$$

$$\text{similarly } f(\Omega_2) = k \ln \Omega_2$$

Q. E. D

④ a) ideal gas  $PV = Nk_B T$  only two independent variables  
i.e.  $E = E(T, V)$ ,  $S = S(T, V)$ , let us go with  $T$  and  $V$

$$dE = T ds - P dV \quad \text{with} \quad ds = \left(\frac{\partial S}{\partial T}\right)_V dT + \left(\frac{\partial S}{\partial V}\right)_T dV$$

$$dE = T \left(\frac{\partial S}{\partial T}\right)_V dT + T \left(\frac{\partial S}{\partial V}\right)_T dV - P dV \quad \dots (1)$$

let us check the dependence of  $E$  on  $V$

$$\left(\frac{\partial E}{\partial V}\right)_T = T \left(\frac{\partial S}{\partial V}\right)_T - P \quad \dots (2)$$

$$\text{but } \left(\frac{\partial S}{\partial V}\right)_T = \frac{\partial(S, T)}{\partial(T, V)} = \frac{\partial(P, V)}{\partial(T, V)} = \left(\frac{\partial P}{\partial T}\right)_V$$

$$\text{so } \left(\frac{\partial E}{\partial V}\right)_T = T \left(\frac{\partial P}{\partial T}\right)_V - P \quad \dots (3)$$

$$\text{for an ideal gas } PV = Nk_B T \Rightarrow P = \frac{Nk_B T}{V}$$

$$\left(\frac{\partial P}{\partial T}\right)_V = \frac{Nk}{V} = \frac{P}{T}$$

$$\Rightarrow \left(\frac{\partial E}{\partial V}\right)_T = T \frac{P}{T} - P$$

$$= P - P = 0$$

so  $E$  does not depend on  $V$

i.e.  $E = E(T)$  only Q.E.D

b) if  $E = E(T)$ , then  $\left(\frac{\partial E}{\partial V}\right)_T = 0$

i.e.  $T \left(\frac{\partial P}{\partial T}\right)_V - P = 0$  or  $P = T \left(\frac{\partial P}{\partial T}\right)_V$

$= T \underbrace{f(V)}$   
 any function of  $V$   
 for ideal gas  
 $f(V) = \frac{Nk_B}{V}$

c) van der Waals eq<sup>n</sup> of state is

$$\left[ P - a \left(\frac{N}{V}\right)^2 \right] (V - Nb) = Nk_B T$$

$\Rightarrow$  solve for  $P$

$$P = \frac{Nk_B T}{(V - Nb)} + a \left(\frac{N}{V}\right)^2$$

now  $C_V = \left(\frac{\partial E}{\partial T}\right)_V$

$$\Rightarrow \left(\frac{\partial C_V}{\partial V}\right)_T = \frac{\partial}{\partial V} \left(\frac{\partial E}{\partial T}\right)_V = \frac{\partial}{\partial T} \left(\frac{\partial E}{\partial V}\right)$$

$$= \frac{\partial}{\partial T} \left[ T \left(\frac{\partial P}{\partial T}\right)_V - P \right]$$

$$= \cancel{\left(\frac{\partial P}{\partial T}\right)_V} + T \left(\frac{\partial^2 P}{\partial T^2}\right)_V - \cancel{\left(\frac{\partial P}{\partial T}\right)_V}$$

$$= T \left(\frac{\partial^2 P}{\partial T^2}\right)_V \quad ; \quad \text{as } \left(\frac{\partial P}{\partial T}\right)_V = \frac{Nk}{V}$$

$$= 0$$

$$\left(\frac{\partial^2 P}{\partial T^2}\right)_V = 0$$

so  $C_V = C_V(T)$

5)  $N = 3$  distinguishable particles

a)  $E = 2\varepsilon$

$3\varepsilon$  ———  $2 \times (1\varepsilon) + 1 \times (0\varepsilon)$

$\varepsilon$  —  $00$

$0$  —  $0$

$$\Omega = \frac{3!}{2! 1!} = \frac{3 \times 2}{2 \times 1 \times 1} = 3 \Rightarrow S = k_B \ln 3$$

b)  $E = 3\varepsilon$

$3\varepsilon$  —  $0$   $1 \times (3\varepsilon) + 2 \times (0\varepsilon)$

$\varepsilon$  — — —

$0$  —  $00$

$$\Omega_1 = \frac{3!}{2! 1!} = 3$$

or  $3\varepsilon$  — — —  $3 \times (1\varepsilon) + 0 \times (3\varepsilon) + 0 \times (3\varepsilon)$

$\varepsilon$  —  $000$

$0$  — — —

$$\Omega_2 = \frac{3!}{2! 0!} = 1$$

$$\Omega_{\text{tot}} = \Omega_1 + \Omega_2 = 3 + 1 = 4 \Rightarrow$$

$$S = k_B \ln 4$$

c)  $E = 9\varepsilon$

$3\varepsilon$  —  $000$

$\varepsilon$  — — —

$0$  — — —

$3 \times (3\varepsilon) + 0 \times (1\varepsilon) + 0 \times (0\varepsilon)$

$$\Omega = \frac{3!}{3! 0!} = 1$$

$$\Rightarrow S = k_B \ln 1 = 0$$

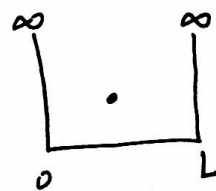
$S = 0$  as expected as there is only one unique configuration

⑥ particle in a 1D box

$$\frac{d^2 \phi(x)}{dx^2} + k^2 \phi(x) = 0 \quad ; \quad k^2 = \frac{2mE}{\hbar^2} \quad \text{and } \phi(0) = \phi(L) = 0$$

$$\phi(x) = A \cos kx + B \sin kx$$

$$\text{but } \phi(0) = 0 \Rightarrow A = 0$$



$$\Rightarrow \phi(x) = B \sin kx$$

$$\text{and } \phi(L) = 0 \Rightarrow \sin kL = 0, \quad kL = n\pi \Rightarrow k_n = \frac{n\pi}{L}$$

$$n = 1, 2, 3, \dots$$

$$\Rightarrow E_n = \frac{\hbar^2 k^2}{2m} = \frac{\hbar^2 \pi^2 n^2}{2mL^2}$$

B can be found from normalization  $\int_0^L |\phi(x)|^2 dx = 1$

$$B^2 \int_0^L \sin^2\left(\frac{n\pi x}{L}\right) dx = 1 \Rightarrow B^2 \frac{L}{n\pi} \int_0^{n\pi} \sin^2 y dy = 1$$

$$B^2 \frac{L}{n\pi} \frac{n\pi}{2} = 1 \Rightarrow B = \sqrt{\frac{2}{L}}$$

$$\Rightarrow \phi_n(x) = \sqrt{\frac{2}{L}} \sin(k_n x)$$

for 3 - non interacting particles,  $E = 38 E_0 = 38 \frac{\hbar^2 \pi^2}{2mL^2}$

$$E = \sum_{i=1}^3 E_i = \sum_{i=1}^3 E_0 n_i^2 = E_0 n_1^2 + E_0 n_2^2 + E_0 n_3^2$$

$$= E_0 (n_1^2 + n_2^2 + n_3^2) = 38 E_0$$

$$\Rightarrow n_1^2 + n_2^2 + n_3^2 = 38$$

two possibilities (5, 3, 2) or (6, 1, 1)



a) 3 distinguishable particles

- (5, 3, 2)  $\Rightarrow$  3! microstates =  $3 \times 2 \times 1 = 6$  microstates

<u>5</u>	<u>3</u>	<u>3</u>	<u>2</u>	<u>2</u>	<u>5</u>
<u>3</u>	<u>5</u>	<u>2</u>	<u>3</u>	<u>5</u>	<u>2</u>
<u>2</u>	<u>2</u>	<u>5</u>	<u>5</u>	<u>3</u>	<u>3</u>
(2,3,5)	(2,5,3)	(5,2,3)	(5,3,2)	(3,5,2)	(3,2,5)

- (6, 1, 1)  $\Rightarrow$   $\frac{3!}{2! 1!} = 3$  microstates (6,1,1), (1,6,1), (1,1,6)

let us label our 3 particles by 1,2,3  $\Rightarrow$   $n=6$   $\frac{3}{1,2}$   $\frac{2}{1,3}$   $\frac{1}{3,2}$   
 such that the state  $n=1$  is doubly occupied.

$\Rightarrow$  total # of microstates =  $6 + 3 = 9 = \Omega$   
 $\Rightarrow S = k_B \ln 9$

b) 3 indistinguishable bosons  
 the 6 microstates of the (5,3,2) are counted now as one microstate as they are identical. similarly the 3 microstates of the (6,1,1) are counted one microstate for the same reason

$\Rightarrow \Omega_{\text{tot}} = 1 + 1 = 2 \Rightarrow S = k_B \ln 2$

c) 3 indistinguishable spin  $1/2$  fermions  
 Assuming that each energy level has  $(2s+1)$  degeneracy, for spin  $1/2$ , we have two particles at most for each energy level (one  $\uparrow$  and one  $\downarrow$ )

- (5, 3, 2) , we have  $2^3 = 8$  microstates

5	<u>↑</u>	<u>↓</u>	<u>↓</u>	<u>↓</u>	<u>↓</u>	<u>↑</u>	<u>↑</u>	<u>↑</u>
3	<u>↑</u>	<u>↑</u>	<u>↓</u>	<u>↓</u>	<u>↑</u>	<u>↑</u>	<u>↓</u>	<u>↓</u>
2	<u>↑</u>	<u>↑</u>	<u>↑</u>	<u>↓</u>	<u>↓</u>	<u>↓</u>	<u>↑</u>	<u>↓</u>

- (6, 1, 1) , here the state with  $n_2 = n_3 = 1$  is doubly occupied , so we have only two microstates

6	<u>↑</u>	<u>↓</u>
1	<u>↑↓</u>	<u>↑↓</u>

$$\Rightarrow \Omega_{\text{tot}} = 8 + 2 = 10$$

$$\Rightarrow \underline{S = k_B \ln 10}$$